phenomena will be discovered in these nanomaterials.

Y. Rosenfeld Hacohen*, E. Grunbaum*, R. Tenne*, J. Sloan†, J. L. Hutchison†

*Department of Materials and Interfaces, Weizmann Institute, Rehovot 76100, Israel †Department of Materials, The University of Oxford, South Parks Road, Oxford OX1 3PH, UK e-mail: cpreshef@weizmann.weizmann.ac.il

- Tenne, R., Margulis, L., Genut, M. & Hodes, G. Nature 360, 444–445 (1992).
- Lindgard, P. A., Birgeneau, R. J., Als-Nielsen, J. & Guggenheim, H. J. J. Phys. C 8, 1059–1069 (1975).
- Ackerman, J., Fouassier, C., Holt, E. M. & Holt, S. L. Inorg. Chem. 11, 3118–3122 (1972).
- 4. Stadelman, P. A. Ultramicroscopy 21, 131-145 (1987).
- 5. Whittaker, E. J. W. Acta Crystallogr. 9, 855-862 (1956).

No fractals in fossil extinction statistics

Statistical analyses of the fossil record seek to discover the mechanisms controlling biotic diversity throughout the Earth's history. Solé *et al.*¹ reported that many extinction time series are statistically self-similar, with 1/f power spectra, suggesting that extinctions are driven by self-organized criticality or by other scale-free internal dynamics of the biosphere. Here we show that the apparent self-similarity and 1/f scaling reported by Solé *et al.* are artefacts of their interpolation methods. Extinction records that are not interpolated show no evidence of fractal scaling.

Self-similarity and 1/f scaling would imply that extinction rates are correlated through time. The problem is that similar correlations are created by Solé et al.'s interpolation techniques. Their raw data are various extinction metrics² for 77 stratigraphic stages that range up to 34 million years (Myr) in length. They interpolate between these 77 points every million years, creating time series with 570 points, 86% of which are interpolations rather than real data. The Fourier power spectra and root-meansquare amplitude spectra of the interpolated time series follow power laws with exponents close to one (Fig. 1a,b), markedly different from the exponents expected from uncorrelated, 'white noise' data sets, leading Solé et al. to infer fractal scaling and attribute it to scale-free dynamics of the biosphere.

But uncorrelated white noise is the wrong null hypothesis in this case, because interpolation introduces points that are correlated with one another. A more appropriate null hypothesis is random numbers substituted for each of the 77 real extinction data, associated with the same stratigraphic stages, and subjected to the same interpolation used in the original analysis. We

evaluated this null hypothesis several thousand times (Fig. 1a,b, red curves), and obtained similar results when we applied the same interpolation and analysis methods to the real extinction data² used by Solé *et al.* (Fig. 1a,b, blue histograms). Random numbers without interpolation yielded the expected 'white noise' spectral exponents (Fig. 1a,b, black curves). These results show that Solé *et al.*'s interpolation methods can generate the fractal scaling they observed.

Solé *et al.* recognized the possibility of interpolation artefacts, but they also found 1/f scaling in ammonoid³ and planktonic foram⁴ diversity fluctuations, with no interpolation (Fig. 1c,d). Their results are incorrect; the actual spectra bear no resemblance

to 1/f spectra (Fig. 1e,f). Solé et al. may have calculated their spectra from time series of taxon counts rather than extinctions or originations. Taxon counts are integrals of the origination–extinction record, so they will be correlated through time and their spectra will be dominated by long-wavelength variations, whatever the dynamics of extinctions and originations.

None of our results suggests fractal scaling in the extinction record or supports recent models of biotic evolution driven by self-organized criticality^{5,6}, which predict that such fractal scaling should occur¹.

The fossil record poses special problems⁷ for conventional spectral methods⁸, which normally require evenly sampled

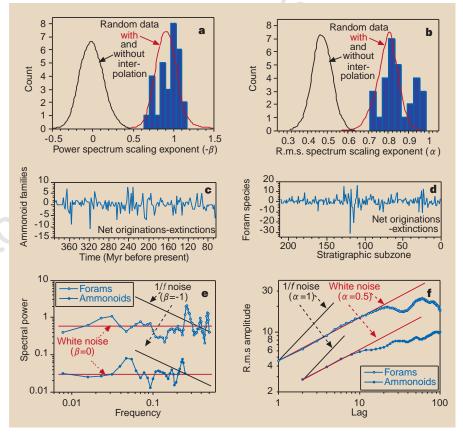


Figure 1 Data analysis. Histograms of fractal scaling exponents for Fourier power spectra (a) and r.m.s. fluctuation spectra (b) for the five extinction measures and six fossil records analysed in ref. 1. Solé et al.'s methods yield scaling exponents close to 1.0, suggesting 1/f scaling. The same interpolation and analysis procedures yield similar scaling exponents (red curves) when random numbers are substituted for the real extinction data at each stratigraphic stage boundary. Linear and step-function interpolations give similar results, as do random numbers from uniform, gaussian or log-normal distributions. Random uncorrelated data, without interpolation, give 'white noise' spectra (black curves); small deviations from white noise scaling arise because the data series are short. Time series of ammonoid3 and planktonic foram4 diversity fluctuations (c, d) yield power spectra (e) and r.m.s. amplitude spectra (f) that deviate markedly from 1/f noise (black lines), and are more consistent with white noise (red lines). The foram data4 (cited incorrectly in ref. 1) are observations at irregularly spaced stratigraphic subzones, not fixed time intervals. Thus the foram spectra cannot be interpreted conventionally, but we show them because Solé et al. report that they obey power laws. Power spectra were calculated from the Fourier transform (with hanning windowing) of the autocorrelation function for lags from 1 to 256 Myr on detrended time series; scaling exponents were estimated by least squares for $0.01 \le f \le 0.2 \,\mathrm{Myr}^{-1}$. Scaling exponents of r.m.s. spectra were calculated for lags from 1 to 30 Myr. Different procedures yield slightly different results, but in each case the results for the real and random interpolated data series are similar. The exponents reported in Table 1 of ref. 1 are more tightly clustered than the exponent histograms shown in a and b here, owing to errors in algorithms in ref. 1 (R.V. Solé, personal communication).

scientific correspondence

data from statistically stationary processes where the record is much longer than the longest wavelength of interest. Most fossil time series violate all three constraints. The fossil record is our only possible data source for studying the long-term dynamics of Earth's biotic system, but such studies require carefully designed hypothesis tests.

James W. Kirchner*, Anne Weil†

*Department of Geology and Geophysics, University of California, Berkeley, California 94720, USA e-mail: kirchner@geomorph.berkeley.edu †Department of Integrative Biology and University of California Museum of Paleontology, University of California, Berkeley, California 94720, USA

- Solé, R. V., Manrubia, S. C., Benton, M. & Bak, P. Nature 388, 764–767 (1997).
- Benton, M. J. (ed.) The Fossil Record 2 (Chapman & Hall, London, 1993).
- 3. House, M. R. Phil. Trans. R. Soc. Lond. B 325, 307–326 (1989).
- 4. Patterson, R. T. & Fowler, A. D. Geology 24, 215–218 (1996).
- 5. Bak, P. & Sneppen, K. Phys. Rev. Lett. 71, 4083–4086 (1993).
- 6. Solé, R. V. & Manrubia, S. C. Phys. Rev. E 54, R42-R45 (1996).
- Connor, E. F. in Patterns and Processes in the History of Life (eds Raup, D. M. & Jablonski, D.) 119–147 (Springer, Berlin, 1986).
- Blackman, R. B. & Tukey, J. W. The Measurement of Power Spectra from the Point of View of Communications Engineering (Dover, New York, 1958).

Arsenic poisoning of Bangladesh groundwater

In Bangladesh and West Bengal, alluvial Ganges aquifers used for public water supply are polluted with naturally occurring arsenic, which adversely affects the health of millions of people. Here we show that the arsenic derives from the reductive dissolution of arsenic-rich iron oxyhydroxides, which in turn are derived from weathering of base-metal sulphides. This finding means it should now be possible, by sedimentological study of the Ganges alluvial sediments, to guide the placement of new water wells so they will be free of arsenic.

As many as a million water wells drilled into Ganges alluvial deposits in Bangladesh and West Bengal may be contaminated with arsenic¹⁻⁶. Measured arsenic concentrations¹⁻⁶ reach up to 1,000 µg l⁻¹, which is above the limit set for drinking water in Bangladesh (50 µg l⁻¹) or that recommended by the World Health Organization (10 µg l⁻¹). Consumption of this contaminated water has led to widespread death and disease¹⁻⁶.

Arsenic has been reported to derive from the oxidation of arsenic-rich pyrite in the aquifer sediments as atmospheric oxygen invades the aquifer in response to a lowering of the water level by abstraction^{4,5}. However, this explanation is not consistent with the following observations³, made on

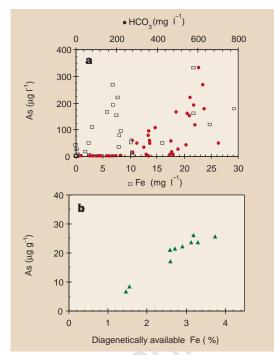


Figure 1 Arsenic chemistry of ground-water and aquifer sediment from Bangladesh. **a,** Arsenic correlates with dissolved iron and with bicarbonate concentrations above 300 mg per litre; the latter is a by-product of iron oxyhydroxide reduction. **b,** Relation of arsenic and iron in aquifer sediments; arsenic concentrations are up to 15 times higher than crustal abundance.

46 wells, typical of those in Bangladesh, that were sampled during May and June of 1997: in oxic (shallow) wells, arsenic concentrations are mostly below 50 µg l-1; in anoxic waters, arsenic concentrations (≤260 µg l⁻¹) correlate with concentrations of dissolved iron (≤29 mg l⁻¹) and bicarbonate (Fig. 1a); and arsenic concentrations increase with depth in wells at Manikganj, Faridpur and Tungipara. These observations suggest that arsenic is released when arsenic-rich iron oxyhydroxides are reduced in anoxic groundwater⁶, a process that solubilizes iron and its absorbed load and increases bicarbonate concentration. Sedimentary iron oxyhydroxides are known to scavenge arsenic⁷ and, in Ganges aquifer sediments, concentrations of diagenetically available iron (≤3.7%) and arsenic (≤26 p.p.m.) correlate well³ (Fig. 1b).

The arsenic-rich groundwater is mostly restricted to the alluvial aquifers of the Ganges delta^{3,6}. The source of arsenic-rich iron oxyhydroxides must therefore lie in the Ganges source region upstream of Bangladesh. Weathered base-metal deposits are known to occur^{6,8–10} in the Ganges basin (at Bihar, Uttar Pradesh, West Bengal), so weathering of these arsenic-rich base-metal sulphides must have supplied arsenic-rich iron oxyhydroxide to downstream Ganges sediments during Late Pleistocene-Recent times. The arsenic-rich iron oxyhydroxides are now being reduced, causing the present problem. Reduction is driven by concentrations of sedimentary organic matter³ of up to 6%.

A knowledge of the sedimentary architecture and distribution of iron, arsenic and reductant carbon in Ganges alluvial sediments will allow the development of a predictive model to guide future aquifer

development in a way that minimizes arsenic pollution. Furthermore, as dissolved iron is oxidized it precipitates as iron oxyhydroxide, which scavenges arsenic from solution. It follows that simple aeration of anoxic Bangladesh groundwater, followed by settling, should remove a considerable amount of arsenic from solution. This simple treatment could be performed on a household or village scale. Although the disposal of the arsenic-rich iron oxyhydroxides would require special arrangement, this would be preferable to either the widespread poisoning that now exists or a return to the use of contaminated surface water for public consumption.

Ross Nickson*, John McArthur*, William Burgess*, Kazi Matin Ahmed†, Peter Ravenscroft‡, Mizanur Rahman§

*Geological Sciences,
University College London,
Gower Street, London WC1E 6BT, UK
†Department of Geology,
University of Dhaka,
Dhaka 1000, Bangladesh
‡Mott MacDonald International Ltd,
122 Gulshan Avenue, Dhaka-1212, Bangladesh
\$GWCII, Bangladesh Water Development Board,
72 Green Road, Dhaka 1207, Bangladesh

- 1. Pearce, F. The Guardian (19 $\&\,26$ February, 1997).
- 2. Dhar, R. K. et al. Curr. Sci. 73, 48-59 (1997).
- Nickson, R. Thesis, University College London (1997).
- Das, D, Basu, G., Chowdhury, T. R. & Chakraborty, D. in Proc. Int. Conf. Arsenic in Groundwater (Calcutta, 1995).
- Saha, A. K. & Chakrabarti, C. in Proc. Int. Conf. Arsenic in Groundwater (Calcutta, 1995).
- Bhattacharya, P., Chattargee, D. & Jacks, G. Water Res. Dev. 13, 79–92 (1997).
- Mok, W. M. & Wai, C.M. Arsenic in the Environment (ed. Nriagu, J. O.) 99–117 (Wiley, New York, 1994).
 Ghosh, S. & De, S. Ind. I. Earth Sci. 22, 183–189 (1995).
- Bannerjee, D. K. Mineral Resources of India (World Press, Calcutta, 1992)
- 10. Wadia, D. N. Geology of India, 4th edn (Tata, 1975).