MINERAL PHYSICS 4: MANTLE RHEOLOGY

RHEOLOGY



- Study of the flow of matter (plastic deformation)
- In geoscience flow of rocks
 - what is the stress strain-rate response?
 - How strong are different rocks
 - Microstructure/texture development during deformation

Marble deformed in torsion (L. Burlini)

WHY CARE ABOUT DEFORMATION BEHAVIOR IN THE DEEP EARTH?



What are the controls on slab subduction

Geochemical reservoirs

Do things from the very deep earth mix with the upper mantle

Springs, dashpots, and sliding blocks

CONSTITUTIVE MODELS

ELASTIC VS PLASTIC DEFORMATION



Elastic – perturbation of atoms from potential well

Non-Elastic – permanent of atoms i.e. moving from one well to another

LINEAR ELASTICITY



 $\sigma_n = E \varepsilon_n \qquad \sigma_s = 2G \varepsilon_s$

CONSTITUTIVE MODELS AND FLOW

Linear (Newtonian) viscous model



- Non-recoverable strain material deforms in response to σ, but if we remove σ we don't return to initial state
- **Cannot** relate σ to ε
- Instead

$$\Delta \sigma_n = 2\mu \dot{\varepsilon}_n$$

 $\sigma_s = 2\mu \dot{\varepsilon}_s$

Twiss and Moores 1992

CONSTITUTIVE MODELS AND FLOW

Rigid/Perfectly Plastic model



- Below yield stress no strain
- Above yield stress plastic unrecoverable strain
- Stress does not exceed yield strength except during acceleration of deformation

CONSTITUTIVE MODELS AND FLOW

Elasto-plastic (Prandtl) model



- Below yield stress elastic recoverable strain
- Above yield stress plastic unrecoverable strain
- Stress does not exceed yield strength except during acceleration of deformation
- Many crystal behave this way

Twiss and Moores 1992

BRITTLE VS DUCTILE

- High confining pressure tends to suppress brittle failure
 - Pressure of brittle ductile transition depends on rock type
- High temperature suppresses brittle failure





Paterson (1958)

CONSTITUTIVE MODELS AND FLOW: MANY OTHERS

Maxwell model

Twiss and Moores 1992

Burgers Model

Kelvin-Voigt Model

Twiss and Moores 1992

MICROSCOPIC MECHANISMS OF DEFORMATION 1: DIFFUSION

DIFFUSION

Fick's 1st law

$$J = -D \frac{\partial c}{\partial x}$$
$$J \equiv \text{flux of atoms} \left(\frac{moles}{m^2 \cdot s}\right)$$
$$D \equiv \text{Diffusion coefficient } (m^2/s)$$
$$c \equiv \text{concentration atoms per unit volume} \left(\frac{moles}{m^3}\right)$$

In 3 D

 $J = -D\nabla c$

For non-steady state

Again in 3 D

$$-DVC$$

a2a

$$\frac{\partial c}{\partial t} = D \frac{\partial^2 c}{\partial x^2}$$
$$\frac{\partial c}{\partial x} = D \nabla^2 c$$

 ∂t

NOTE: in the above equations D is assumed to be isotropic. In general it is not! It is a 2nd rank tensor D_{ii}

DIFFUSION LENGTH SCALE

$$D \approx \frac{l^2}{t}$$

In minerals D can range from something like 10⁻¹⁰ m²/s for H diffusion to something like 10⁻²¹ m²/s for U-Th

- So H can move 1m in ~300 years very fast
- U-Th can move ~1mm over the age of the Earth very slow

In general diffusion length scales are small, i.e. inefficient way to homogenize.

DIFFUSION IN CRYSTALS: THE ROLE OF DEFECTS

Perfect crystal

Vacancy Point Defect

Mühlbacher 2015

 $D = \frac{1}{6}a^2\Gamma$

 $a \equiv length scale of a jump$ $\Gamma \equiv frequency of a jump (probablity)$

We are dealing with the probability to jump from one atomic site to another

- Very difficult in a perfect crystal
- Diffusion is related to defect concentration

 $\Gamma = \Gamma_{f} \Gamma_{m}$

 $\Gamma_f \equiv defect \ fraction$ (probablity of finding a defect)

> $\Gamma_m \equiv probability \ to \ jump$ when a defect is present

PRESSURE AND TEMPERATURE DEPENDENCE

- Both defect fraction and jump probability have an Arrhenius type temperature dependence
- Pressure tends to decrease vacancies and

$$D = D_0 exp\left(-\frac{E_a + PV_a}{RT}\right)$$

 $D_0 \equiv frequency factor$

 $R \equiv gas \ constant = 8.314510 \frac{\mathsf{J}}{mol \cdot K}$

 $E_a \equiv activation \ energy$

 $V_a \equiv activation \ volume$ -takes into account the effects of pressure

HOMOLOGOUS TEMPERATURE SCALING

$$D(P,T) = D_0 \exp\left(-\frac{gT_m(P)}{T}\right)$$

 $g = H^*(P)/RT_m(P)$

 $T_m(P) \equiv$ Melting temperature at a given pressure

 $g \equiv$ dimensionless constant determined empirically

Liu et al 2006

 $H^*(P) \equiv$ activation enthalpy at a given pressure $R \equiv$ gas constant

FLOW LAWS 1 : DIFFUSION CREEP

DIFFUSION CREEP

- Newtonian (linear) rheology $\dot{\varepsilon} \propto \sigma$
- Intracrystalline (Nabbaro-Herring) or along grain boundaries (Coble)
- Efficiency depends on diffusion coefficients and grain size
- Dominates at low stress and small grain size
 - Iow strain rate and high temperature

NABARRO (1948), HERRING (1950): DIFFUSION CREEP

- Non-hydrostatic stress causes differences in vacancy concentrations of crystal surfaces
- Vacancies move from low compressive stress to high compressive stress and atoms move from high compressive stress to low

 $J = -D_{v}\nabla c$ $D_{v} \equiv diffusion \ coefficent \ of \ vacancies$

$$J \approx \alpha D_v \frac{c_v^+ - c_v^-}{h}$$

 $\alpha = geometric \ factor - (grain \ shape \ and \ diffusion \ path)$ $h = diffusion \ length \ scale \ (grain \ size)$

NABARRO (1948), HERRING (1950): DIFFUSION CREEP

So our change in length per second is

$$\frac{\Delta h}{dt} = \frac{\phi \Omega}{h^2}$$

$$\phi \equiv number of vacancies / sec = Jh^2$$

$$\Omega \equiv \text{atomic volume}$$

And strain rate ends up looking like

$$\dot{\varepsilon} = \alpha D_v \frac{c_v^+ - c_v^-}{h} \frac{\Omega}{h}$$

Since vacancy concentration has an Arrenius type relationship

$$\dot{\varepsilon} \approx \alpha \frac{D_{\nu} c_{\nu 0} \Omega}{h^2} \left(\frac{\sigma \Omega}{RT} \right)$$
$$\sigma \equiv \sigma_1 - \sigma_3$$

Relating the diffusion coefficient for self-diffusion to vacancy diffusion we end up with

$$\dot{\varepsilon} = \frac{\alpha D \sigma \Omega}{h^2 R T}$$

If GBS
$$\rightarrow$$
impossible $\alpha = 16$
If GBS \rightarrow possible $\alpha = 40$

COBLE CREEP

- Assumes grains are spherical and GBS is possible
- Diffusion creep where diffusion occurs along grain boundaries

$$\dot{\varepsilon} = \frac{\alpha D_{gb} \delta \sigma \Omega}{\pi h^3 RT}$$

 $D_{gb} \equiv diffusion \ coefficient \ of \ grain \ boundary$

 $\delta \equiv thickness \ of \ grain \ boundary, typically \sim atomic \ length \ scale \\ \alpha = 148$

Note: N-H and Coble really happen in tandem

MICROSCOPIC MECHANISMS OF DEFORMATION 2: DISLOCATIONS

THEORETICAL SHEAR STRENGTH VS EXPERIMENTAL

• Frenkel 1926

- ideal shear strength of a crystal

$$\tau_m = \frac{G}{2\pi} \frac{b}{a}$$

 $G \equiv shear modulus$

• if $b \sim a$ then

$$\tau_m = \frac{G}{2\pi}$$

For metals shear modulus G is in the range of 20-150 GPa so τ_m should be ~3-30 GPa but experiments show more like 0.5-10 MPa!

CONCEPT OF DISLOCATIONS TAYLOR (1934), OROWAN (1934), POLANYI (1934)

Dislocations in dolomite

http://faculty.wiu.edu/K-Hall/CM/Dislocations.htm

SCREW DISLOCATION

Barber et al 2007

MIXED DISLOCATIONS

PEIERLS STRESS/LATTICE FRICTION, PEIERLS (1940)

- Frequently called Peierls-Nabarro model
 - Peierls made algebraic error and had an extra 2 in the exponent
 - Nabarro corrected this in 1947

"All I did was formulate and solve the equations representing his [Orowan] model. I had wanted this to be a joint paper with Orowan, but he refused The matter did not seem important at the time and I did not take more interest in the problem of dislocations"

$$\sigma_p = \frac{2G}{w} exp\left(-\frac{2\pi}{w}\frac{d_{hkl}}{\overline{b}}\right)$$

 $w \rightarrow$ depends on dislocation character w = 1 for screw dislocations w = 1 - v for edge dislocations, where $v \equiv$ Poisson ratio

 $G \equiv shear modulus$ $d_{hkl} \equiv spacing of the glide plane (lattice plane spacing)$

FLOW LAWS 2 :DISLOCATION GLIDE AND CREEP

DEFORMATION MECHANISMS AT HIGH P-T DISLOCATION GLIDE AND CREEP

http://faculty.wiu.edu/K-Hall/CM/Dislocations.htm

- Glide is governed by "lattice friction" Peierls Stress – stress needed to move a dislocation without thermal activation
- Slip system {hkl}<uvw>
- Dislocation creep glide + climb (diffusion)
- Non linear $\dot{\varepsilon} \propto \sigma^n$
 - For silicates and oxides *n* ~3-5

DISLOCATION GLIDE

 Average shear strain in the crystal multiple (N#) of parallel dislocations

$$\boldsymbol{\varepsilon}_{avg} = N \frac{\overline{b}}{d} \frac{\Delta L}{L}$$

And if we deal with one unit of depth we get

•
$$\varepsilon_{avg} = \frac{N}{V} \ \overline{b} \Delta L$$

- Define dislocation density as the number of dislocations per volume $ho = rac{N}{V}$
 - $\varepsilon_{avg} = \overline{b}\rho\Delta L$
- Take the time derivative to get strain rate
- $\dot{arepsilon} = \overline{b}
 ho rac{d \Delta L}{dt}$
- Since $\frac{d\Delta L}{dt}$ is velocity

WHY HAVE CLIMB? DISLOCATION TANGLES: WORK HARDENING

DISLOCATION CREEP

- Dislocation creep glide + climb (diffusion)
- At high temperature climb allows dislocations to climb around obstacles by absorbing or emitting vacancies

Rewrite velocity in terms of time to glide t_g and time to climb t_c

$$\nu = \frac{\Delta L}{t_g + t_c}$$

For climb controlled

$$\nu = \frac{\Delta L}{t_g + t_c} \approx \frac{\Delta L}{t_c} = \Delta L \left(\frac{\nu_c}{d_{hkl}} \right)$$

Then the Orowan looks like

$$\dot{\varepsilon} = \rho \overline{\boldsymbol{b}} v_c \left(\frac{\Delta L}{d_{hkl}} \right)$$

WEERTMAN MODEL(1968, 1972)

Frank Reed Source in Si Barrett, Nix, and Tetelman 1973

- Treat dislocations as loops that propagate out (like Frank Read source)
- Loops that are a distance of d apart can climb into each other and annihilate (recovery) dipole
- Assumes balance of dislocation formation and recovery steady state.

WEERTMAN MODEL(1968, 1972)

Simplifying

$$\dot{\varepsilon} = \frac{4\overline{b}}{(\pi M)^{1/2} (d_{hkl})^{7/2}} v_c$$

the density of dislocation sources (*M*) is $M = \frac{1}{\pi \Delta L^2 d_{hkl}}$

assuming that dislocations become mobile only when a dipole annihilates (steady state)

$$\rho = M\left(\frac{4\pi\Delta L^2}{d_{hkl}}\right) = \left(\frac{4}{d_{hkl}^2}\right)$$

Putting this into climb controlled Orowan gives

$$\dot{\varepsilon} = \left(\frac{4}{d_{hkl}^2}\right) \overline{\boldsymbol{b}} \frac{\Delta L}{d_{hkl}} v_c = 4 \overline{\boldsymbol{b}} \frac{\Delta L}{d_{hkl}^3} v_c$$

Climb velocity is to the diffusion coefficient and stress so

$$\nu_c \propto D\sigma$$

and

$$\sigma = G \, \frac{\overline{\boldsymbol{b}}}{d_{hkl}}$$

$$\dot{\varepsilon} = \alpha \frac{D}{\overline{\boldsymbol{b}}^{3.5} M^{0.5}} \left(\frac{\sigma}{G}\right)^{4.5} \frac{G\Omega}{RT}$$

Note: No grain size dependence No single viscosity

GENERAL FLOW LAW: DISLOCATION AND DIFFUSION CREEP

$$\dot{\varepsilon} = A \left(\frac{\sigma}{G}\right)^n \left(\frac{b}{h}\right)^m exp\left(-\frac{E_a + PV_a}{RT}\right)$$
Pre exponential factor – determined empirically
$$b \equiv \text{lattice spacing}$$

 $h \equiv$ grain size

- Diffusion creep n=1 and m=2.5
- For dislocation creep n=3-5 and m=0
- We have $\dot{\varepsilon} \propto \sigma^n vs \ \dot{\varepsilon} \propto \sigma$

 $A \equiv$

- If stress is small then diffusion dominates because $\dot{\varepsilon}$ becomes v.v. small if n>1
- Diffusion time scale increase as the square of length/grain size $(D \approx \frac{l^2}{t})$ so need small grain size
- If temperature is high diffusion become more efficient because the diffusion coefficient increases exponentially
WHERE DO YOU EXPECT DISLOCATION CREEP VS DIFFUSION CREEP



USGS

EFFECTS OF PARAMETERS IN THE FLOW LAW



FLOW LAWS 3: DEFORMATION MECHANISM MAPS

CONSTRUCTION OF DEFORMATION MECHANISMS MAPS



- Semi empirical compile material properties and fit data to flow laws
- In general the mechanisms that yields the fastest strain rate for a given set of conditions dominates
- Field boundaries are where two mechanisms have equal rates

DEFORMATION MECHANISMS MAPS FOR OLIVINE



Frost and Ashby 1982

DISLOCATIONS: TEXTURE AND ANISOTROPY

TEXTURE IN GLACIER SHEETS



University of Copenhagen Center for Ice and Climate, Niels Bohr Institute http://www.iceandclimate.nbi.ku.dk/research/flowofice/ice_crystal_structure/

SCHMID'S LAW







Rosbaud and Schmid 1925

- Resolved shear stress
 - $\tau_{RSS} = \sigma(\cos\phi\cos\lambda)$
- Schmid factor $\equiv \cos \phi \cos \lambda$

• Glide occurs when $\tau_{RSS} \ge \tau_{CRSS}$

MECHANISMS OF TEXTURE DEVELOPMENT





- Texture development
 - Twinning
 - Dislocation glide (hkl)[uvw]



Taylor : iso-strain

- Stresses too high
- Compatibility
 - Need five slip systems



Sachs : iso-stress

- Stresses too low
- Only favorably oriented slip systems deform

ANISOTROPY IN THE UPPER MANTLE

- Seismic waves (P) travel faster along shear zones than across them
 - Raitt 1963: Shor 1964
- Hess 1964 anisotropy due to texture (CPO)
- Raleigh 1969
 - Deformation of olivine is controlled by slip on {0kl}[100]
- Francis 1969
 - [100] aligns with flow direction
 - Fast P-wave propagation is flow direction







ACTIVITY OF MULTIPLE SLIP SYSTEMS



It is very difficult to activate only one slip system

REPRESENTATION OF TEXTURE: POLE FIGURES





TEXTURES: INVERSE POLE FIGURES



- Probability of finding a pole to a lattice plane in the compression direction
- multiples of random distribution (m.r.d.)
- IPF has symmetry of mineral

A CAUTION ON INTERPRETATIONS OF TEXTURES



TECHNIQUES TO STUDY MANTLE RHEOLOGY (TRANSITION ZONE AND LOWER MANTLE)

WHAT DO WE WANT?

- Controlled rheology experiments
 - At P-T conditions of the core mantle boundary
 - Over a range of compositions
 - And a range of stresses/strain rates



Something like this that goes to ~130 GPa and ~3000K...

WHAT CAN WE DO?

Controlled rheology devices

- RDA (Yamazaki and Karato 2001), D-DIA (Wang et al 2003), DT-Cup (Hunt et al 2014)
 - Controlled strain rates
 - *ċ* ~ 10⁻⁵ s⁻¹
 - Relatively large samples
 - Better control of sample environment
- P-T conditions of transition zone and top of the lower mantle

Diamond anvil cell

- Radial diffraction in the Diamond anvil cell
 - Poorly constrained strain/strain rate
 - $\dot{\varepsilon} \sim 10^{-3} \, \mathrm{s}^{-1} \, \mathrm{to} \, 10^{-4} \, \mathrm{s}^{-1}$
 - But typically discontinuous
 - Small samples
 - Relatively poor control of sample
 environment

- Pressures of inner core but ambient T
- ~60 GPa and 1500-2000 K (Miyagi et al 2013)

DEFORMATION-DIA (D-DIA)



- P-T condition equivalent transition zone and top of lower mantle
- 1-2mm samples





http://www.compres.stonybrook.edu

RADIOGRAPHY-MEASURE STRAIN



ROTATIONAL DRICKAMER APPARATUS – SHEAR STRAIN





Karato and Weidner 2008

P-T conditions of base of transition zone/ top of lower mantle ~660 km

STRAIN RATE AND STRESS STRAIN

Ringwoodite at 23 GPa and 1800 K



Miyagi et al. 2014

DIAMOND ANVIL CELL (DAC)



- Trade off between sample size and maximum pressure
- Pressure > than center of the earth (365 GPa)
- No good way to measure strain

RESISTIVE HEATING









P-T condition ~1500km depth

THEORETICAL STUDIES ON MANTLE RHEOLOGY



Kraych et al 2016

WHAT DO WE DO IN THE LAB VS NATURE



DEFORMATION MECHANISMS IN THE EARTH

SO HOW DO WE KNOW WHAT MECHANISMS OPERATE IN THE EARTH?

Upper Mantle

- Microstructural observations of rocks brought to the surface
 - Texture (crystallographic preferred orientation) indicates dislocation creep (glide)
 - Large strain but no texture most likely diffusion creep
 - Note rocks that are brought to the surface have complex deformation histories
- Seismic anisotropy most likely due to texture from dislocation creep



Park and Jung 2017

Transition Zone and Lower Mantle

- No samples for microstructure
- Rely on seismic observations

ANISOTROPY IN THE DEEP EARTH



STRENGTH OF MANTLE MINERALS

WADSLEYITE: EXPERIMENTS AND THEORY



RINGWOODITE: EXPERIMENTS AND THEORY



BRIDGMANITE: EXPERIMENTS AND THEORY

[100](010)

1500

T (K)

P = 30 GPa; $\dot{\varepsilon} = 10^{-5} \text{ s}^{-1}$; $\rho = 10^{13} \text{ m}^{-2}$

[010](100)

1000

 $\frac{1}{2}\langle 110\rangle \{100\}$ $\frac{1}{2}\langle 110\rangle \{110\}$

500

14

12

10

8

6

4

2

0

0

Bg

MgO

Engineering stress (GPa)



T (K)

Kraych et al 2017

2000

BRIDGMANITE AND FERROPERICLASE



BRIDGMANITE AND POST-PEROVSKITE



Studies of low pressure analog materials seem to indicate that pPv is much weaker than Brg

Hunt et al (2009) CalrO₃; Dobson et al (2012) NaCoF₃

COMPLICATIONS: MULTIPLE PHASES

VISCOSITY OF 2-PHASE COMPOSITES



Single phase

•
$$\mu = \frac{\sigma}{2\dot{\varepsilon}}$$

2 phase viscosity

- LBF vs IWL microstructure
- Takeda 1998 –formulae for viscosity of LBF and IWL (Newtonian only)


POLY PHASE PLASTICITY



iso-strain (strain rate)

• Stresses too high

iso-stress

Stresses too low

More like LBF

More like IWL

BRIDGMANITE AND FERROPERICLASE



Do these show the same or different strength ratios?

What is the stress-strain compatibility state?

WHERE DO WE EXPECT LBF VS IWL



- LBF
 - Large volume fraction of strong phase
 - Low strain
- IWL
 - Large volume fraction of weak phase
 - Large strain

IMPLICATIONS FOR DEEP EARTH CONVECTION



Wang et al 2011- shear deformation of Olv+Fe,Ni,S alloy Inset d2 - Pv+Fp from Yamazaki et al. 2009

TEXTURES OF TWO PHASE MATERIALS

MgSiO₃ BRIDGMANITE IN THE HIGH-T DAC





Couper et al in prep

TEXTURE AND MICROSTRUCTURE



Kaercher et al. 2016

MICROTOMOGRAPHY AND FFT-VPSC



Kaercher et al. 2016

SOME CONCLUSIONS AND THOUGHTS

- High stress regions should favor dislocation creep
 - Colder, higher strain rates –slab and edges of convection cells
 - Expect seismic anisotropy
 - $\dot{\varepsilon} \propto \sigma^n$
- Low stress regions should favor diffusion creep
 - "stagnant" convection cores, possibly plumes
 - Expect seismic isotropy
 - $\dot{\varepsilon} \propto \sigma$
- High strains tend to favor interconnected weak layer so softer phase (Fp) should dominate viscosity
- Low strains favor load bearing framework so harder phase (Brg) tends to dominate
- Texture in poly-phase aggregates is not straight forward